

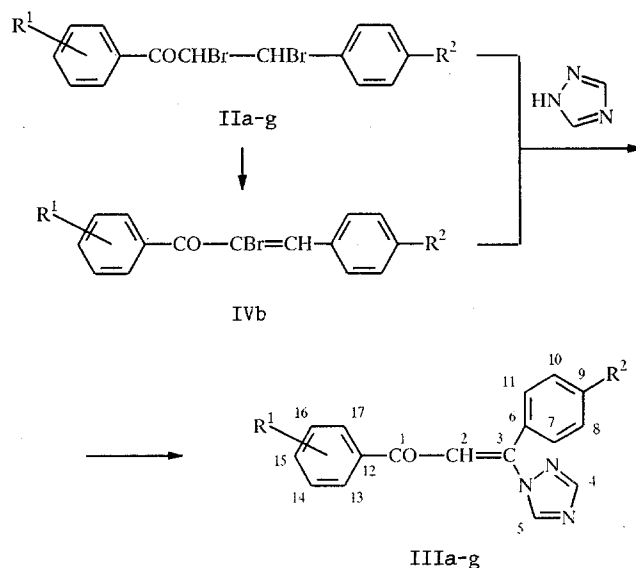
LETTERS TO THE EDITOR

1,3-DIARYL-3-(1H-1,2,4-TRIAZOL-1-YL)PROP-2-EN-1-ONES

M. A. Rekhter, G. N. Grushetskaya, A. A. Panasenko,
and M. Z. Krimer

In recent years research teams have shown much interest in *N*-vinyltriazoles — 1,3-diaryl-2-(1H-1,2,4-triazol-1-yl)prop-2-en-1-ones (I), which show fungicidal and growth-regulating activities [1, 2].

We have, for the first time, prepared representatives of a new type of compound of this class — 1,3-diaryl-3-(1H-1,2,4-triazol-1-yl)prop-2-ene-1-ones (IIIa-g) which are regioisomers of *N*-vinyltriazoles I.



II, IIIa $R^1 = R^2 = H$; b $R^1 = 4\text{-Cl}$, $R^2 = H$; c $R^1 = 4\text{-Cl}$, $R^2 = \text{Cl}$; d $R^1 = 2,4\text{-Cl}_2$, $R^2 = \text{NO}_2$; e $R^1 = H$, $R^2 = \text{OMe}$; f $R^1 = 4\text{-Cl}$, $R^2 = \text{OMe}$; g $R^1 = 4\text{-Cl}$, $R^2 = \text{NO}_2$

As it proved, condensation of chalcone dibromides IIa-g with 1,2,4-triazole in DMSO led exclusively to the formation of compounds IIIa-g in the form of a mixture of *E* and *Z* isomers. From these mixtures the *E* isomers of IIIa-g were isolated by crystallization from methanol. The structure of compounds IIIa-g and their isomeric composition were established by means of ^{13}C NMR spectroscopy and the structure of the *E* isomers of IIIb (mp 177-178°C) and IIIc (mp 157-159°C) were confirmed by x-ray structural studies.

The regiospecific formation of *N*-vinyltriazoles IIIa-g from chalcone dibromides IIa-g is evidence that in contrast to the known reactions of chalcone dibromides [3] the direction of the present reaction is independent of the nature of the radical R^2 and we have confirmed this by preparing, under the conditions described, *N*-vinyltriazole IIIb (80% yield) from the monobromoketone IVb. Further studies will be devoted to clarifying the mechanism of this reaction.

Thus, a solution of 4.82 g (10 mmoles) chalcone dibromide IIc and 2.76 g (40 mmoles) 1,2,4-triazole in 50 ml DMSO was heated 4 h at 110-130°C, cooled to room temperature, poured into 500 ml water and extracted with 4 × 100 ml benzene. The benzene extract was washed with water and dried over Na_2SO_4 . The residue obtained after removing the benzene was

separated chromatographically (Silpearl, 3:2 ethyl acetate–hexane) to give 3.31 g (85%) (E,Z)-3-(4-nitrophenyl)-3-(1H-1,2,4-triazol-1-yl)-1-(2,4-dichlorophenyl)-prop-2-en-1-one (III, C₁₇H₁₀Cl₂N₄O₃). ¹³C NMR spectrum (DMSO-D₆): E isomer (50%): 189.7 (C₍₁₎), 117.7 (C₍₂₎), 144.7 (C₍₃₎), 153.2 (C₍₄₎), 145.7 (C₍₅₎), 136.8 (C₍₆₎), 131.5 (C₍₇₎, C₍₁₁₎), 123.4 (C₍₈₎, C₍₁₀₎), 148.2 (C₍₉₎), 136.6 (C₍₁₂₎), 136.9 (C₍₁₃₎), 129.7 (C₍₁₄₎), 137.8 (C₍₁₅₎), 127.5 (C₍₁₆₎), 131.6 (C₍₁₇₎); Z isomer (50%): 189.0 (C₍₁₎), 124.8 (C₍₂₎), 141.0 (C₍₃₎), 152.6 (C₍₄₎), 146.7 (C₍₅₎), 139.9 (C₍₆₎), 129.1 (C₍₇₎, C₍₁₁₎), 123.9 (C₍₈₎, C₍₁₀₎), 148.6 (C₍₉₎), 132.2 (C₍₁₂₎), 135.3 (C₍₁₃₎), 130.2 (C₍₁₄₎), 136.8 (C₍₁₅₎), 127.3 (C₍₁₆₎), 131.9 (C₍₁₇₎).

The remaining N-vinyltriazoles IIIa-c, e-g were prepared similarly; their yields were 50, 72, 96, 41, 48, and 72%, respectively.

The results of elemental analyses of the compounds prepared corresponded to those calculated.

REFERENCES

1. I. M. Mil'shtein, Zh. Vses. Khim. Ova, **33**, 687 (1988).
2. W. Reiser, W. Draber, K. H. Büchel, K. Lurssen, P.-E. Frohberger, and P. Volker, US Patent 4,749,405 (1989).
3. R. Elderfield, Heterocyclic Compounds, Wiley, New York (1957), p. 185.